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### Three new cassane diterpenes from the seeds of *Caesalpinia minax* Hance

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## ORIGINAL ARTICLE

### Three new cassane diterpenes from the seeds of *Caesalpinia minax* Hance

Zhao-Hua Wu<sup>a</sup>, Jian Huang<sup>bc</sup>, Wei-Dong Li<sup>bc</sup>, Li-Jun Wu<sup>bc</sup> and Hui-Yuan Gao<sup>bc\*</sup>

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Three new cassane-diterpene-lactones, methyl 1 $\alpha$ ,7 $\beta$ -diacetoxy-5 $\alpha$ ,12 $\alpha$ -dihydroxy-cass-13(15)-en-16,12-olide-17 $\beta$ -carboxylate (**1**), methyl 7 $\beta$ -acetoxy-1 $\alpha$ ,5 $\alpha$ ,12 $\alpha$ -trihydroxy-cass-13(15)-en-16,12-olide-17 $\beta$ -carboxylate (**2**), and 12 $\alpha$ -ethoxyl-1 $\alpha$ ,6 $\alpha$ ,7 $\beta$ -triacetoxy-5 $\alpha$ ,14 $\beta$ -dihydroxy-cass-13(15)-en-16,12-olide (**3**), were isolated from the seeds of *Caesalpinia minax* Hance. Their structures were established on the basis of HR-ESI-MS, 1D and 2D NMR spectral analysis.

**Keywords:** *Caesalpinia minax* Hance; Fabaceae; cassane diterpenes

#### 1. Introduction

Plants belonging to the genus *Caesalpinia* (Fabaceae) are widely distributed throughout the tropical and subtropical regions, and many species are used in different systems of traditional medicine for the treatment of many diseases such as antiviral [1–4], anti-inflammatory, anti-analgesic [5], radical growth regulation [6], and anti-tumor [7] agents. In addition, a large number of chemical investigations on this genus have found that diterpene derivatives with a cassane skeleton are the main bioactive components, and much attention has been given to these compounds recently. The seeds of *Caesalpinia minax* Hance are called ‘Ku-Shi-Lian’ in China, as a folk medicine, especially in Guangxi Zhuang Autonomous Region and Sichuan Province, to treat influenza, fever,

and dysentery [8]. For the purpose of finding new anti-flu viral agents from this plant, the chemical study on the seeds led to the isolation of three new cassane-diterpene-lactones. In this paper, we describe the separation and structural elucidation of these new compounds **1–3** (Figure 1).

#### 2. Results and discussion

Compound **1** was obtained as colorless needles (CHCl<sub>3</sub>). Its molecular formula was established as C<sub>25</sub>H<sub>34</sub>O<sub>10</sub> by the positive ion peak at *m/z* 495.2222 [M + H]<sup>+</sup> in HR-ESI-MS. The absorption bands at 3408 and 1744 cm<sup>-1</sup> in the IR spectrum indicated the presence of the hydroxyl and carbonyl groups. The <sup>1</sup>H NMR spectrum (Table 1) exhibited signals due to six methyl singlets [ $\delta_{\text{H}}$  1.03 (6H, s),

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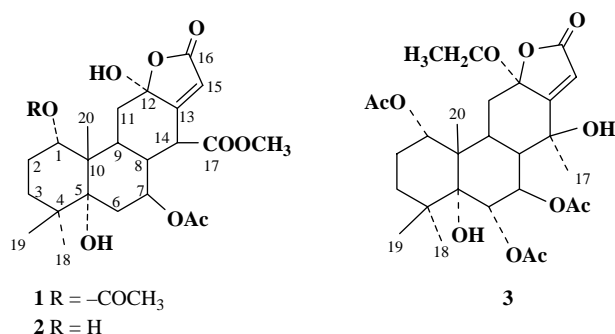


Figure 1. Structures of compounds **1**–**3**.

1.09 (3H, s), 2.17 (3H, s), 2.00 (3H, s), 3.79 (3H, s)], two oxygen-substituted methines [ $\delta_{\text{H}}$  5.23 (1H, td,  $J = 10.8$ , 5.4 Hz), 4.90 (1H, br s)], one olefin proton at  $\delta_{\text{H}}$  5.69 (1H, d,  $J = 1.8$  Hz), along with other alkyl signals. The  $^{13}\text{C}$  NMR spectrum (Table 1) showed 25 carbon signals, including four carbons for two acetoxy groups ( $\delta_{\text{C}}$  21.2, 169.9, 21.3, 169.7), three tertiary methyls ( $\delta_{\text{C}}$  17.5, 24.5, 27.9), one methoxy ( $\delta_{\text{C}}$  52.2), five methylenes ( $\delta_{\text{C}}$  22.6, 29.9, 32.1, 36.0, 36.1), five methines ( $\delta_{\text{C}}$  44.0, 48.4, 74.7, 74.8, 115.4), and seven quaternary carbons ( $\delta_{\text{C}}$  38.3, 43.4, 78.2, 104.1, 164.8, 169.3, 171.1). The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data (Table 1) for **1** suggested that it was a cassane derivative with the  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone moiety [9]. The existence of the  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone moiety could be verified by four carbon signals at  $\delta_{\text{C}}$  104.1, 115.4, 164.8, and 169.3. Moreover, the locations of two acetoxy groups were determined to be at C-1 and C-7, respectively, on the basis of long-range correlations of H-1 ( $\delta_{\text{H}}$  4.90) with one carbonyl at  $\delta_{\text{C}}$  169.7 and H-7 ( $\delta_{\text{H}}$  5.23) with the other at  $\delta_{\text{C}}$  169.9. Compound **1** was considered to be a derivative of neocaesalpin N with two acetoxy groups in the structure, which was confirmed by the analysis of HSQC and HMBC spectra completely (Figure 2) [9]. In addition, compound **1** persisted the same stereochemistry as that of neocaesalpin N,

according to the NOESY correlations (Figure 2) of CH<sub>3</sub>-20/H-1, H-6<sub>ax</sub>, H-8; H-14/H-7, H-9 and the coupling constant for H-14 ( $\delta_{\text{H}}$  3.42, dd,  $J = 10.5$ , 1.8 Hz). Therefore, compound **1** was elucidated as methyl 1 $\alpha$ ,7 $\beta$ -diacetoxy-5 $\alpha$ ,12 $\alpha$ -dihydroxy-cass-13(15)-en-16,12-olide-17 $\beta$ -carboxylate.

Compound **2** was obtained as white needles (CHCl<sub>3</sub>) and its molecular formula was assigned as C<sub>23</sub>H<sub>32</sub>O<sub>9</sub> by the positive ion peak at  $m/z$  475.1941 [M + Na]<sup>+</sup>. The IR spectrum of **2** showed the absorption bands at 3366 (hydroxyl) and 1747 cm<sup>-1</sup> (ester carbonyls). The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra exhibited almost the same signal patterns as those of **1** except for the lack of an acetoxy group [ $\delta_{\text{H}}$  2.17 (3H, s) in the  $^1\text{H}$  NMR spectrum and  $\delta_{\text{C}}$  21.3, 169.7 in the  $^{13}\text{C}$  NMR spectrum of **1**]. Comparing with compound **1**, the chemical shift for C-1 in **2** was upfielded from  $\delta_{\text{C}}$  74.8 to 72.1, which indicated the presence of a hydroxyl at this position, and this hypothesis could be confirmed by its HMBC experiment. The long-range correlations of H-1 ( $\delta_{\text{H}}$  3.66) with C-3 ( $\delta_{\text{C}}$  29.7), C-5 ( $\delta_{\text{C}}$  80.1), and C-10 ( $\delta_{\text{C}}$  43.4) were observed. Thus, compound **2** was elucidated as methyl 7 $\beta$ -acetoxy-1 $\alpha$ ,5 $\alpha$ ,12 $\alpha$ -trihydroxy-cass-13(15)-en-16,12-olide-17 $\beta$ -carboxylate.

Compound **3** was obtained as a white amorphous solid (CHCl<sub>3</sub>). Its molecular formula was determined as C<sub>28</sub>H<sub>40</sub>O<sub>11</sub> by

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR spectral data of compounds **1–3** (CDCl<sub>3</sub>, J in Hz).

No.	1		2		3	
	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$
1	4.90 (br s)	74.8	3.66 (br s)	72.1	4.87 (br s)	75.1
2	1.78 (m) $\alpha$	22.6	1.74 (m) $\alpha$	25.6	1.76 (m) $\alpha$	22.3
3	2.00 (m) $\beta$	29.9	1.99 (m) $\beta$	29.7	1.92 (m) $\beta$	32.2
4	1.15 (m) $\alpha$	38.3	1.13 (m) $\alpha$	38.5	1.13 (m) $\alpha$	38.4
5	1.74 (m) $\beta$	78.2	1.95 (m) $\beta$	80.1	1.76 (m) $\beta$	79.1
6	—	32.1	—	32.9	—	74.7
7	1.56 (dd, 13.2, 10.8) $\beta$	74.7	1.63 (dd, 13.2, 10.8) $\beta$	75.1	5.65 (t, 9.0)	72.6
8	2.18 (dd, 13.2, 5.4) $\alpha$	44.0	2.08 (dd, 13.2, 5.4) $\alpha$	43.7	1.92 (m)	50.8
9	5.23 (td, 10.8, 5.4)	36.0	5.11 (td, 10.8, 5.4)	35.9	2.69 (t, 12.6)	33.5
10	2.24 (m)	43.4	2.28 (m)	43.4	—	44.4
11	2.84 (td, 12.6, 2.4)	36.1	2.97 (td, 12.0, 3.0)	36.1	—	37.6
12	1.44 (dd, 13.2, 12.6) $\beta$	104.1	1.55 (dd, 13.2, 12.6) $\beta$	104.1	1.33 (t, 12.6) $\beta$	106.3
13	2.06 (dd, 13.2, 2.4) $\alpha$	164.8	2.56 (dd, 13.2, 3.0) $\alpha$	164.2	2.12 (d, 10.8) $\alpha$	172.5
14	—	48.4	—	48.2	—	74.9
15	3.42 (dd, 10.5, 1.8)	115.4	3.44 (dd, 10.2, 1.8)	115.4	2.31 (s, 14-OH)	116.4
16	5.69 (d, 1.8)	169.3	5.67 (d, 1.8)	169.3	6.06 (s)	168.6
17	—	171.1	—	171.2	—	20.0
18	1.03 (s)	27.9	0.99 (s)	27.7	1.57 (s)	30.4
19	1.03 (s)	24.5	1.03 (s)	24.6	1.14 (s)	24.4
20	1.09 (s)	17.5	1.02 (s)	17.5	1.13 (s)	16.6
1-OCOCH <sub>3</sub>	2.17 (s)	169.7	—	—	1.15 (s)	168.8
6-OCOCH <sub>3</sub>	—	21.3	—	—	2.14 (s)	21.1
7-OCOCH <sub>3</sub>	—	—	—	—	—	170.2
—	2.00 (s)	169.9	2.01 (s)	170.3	2.08 (s)	21.5
—	—	21.2	—	21.2	2.01 (s)	170.7
—	—	—	—	—	—	21.3

Table 1 – continued

No.	1		2		3	
	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$	$\delta_{\text{H}}$	$\delta_{\text{C}}$
12-OCH <sub>2</sub> CH <sub>3</sub>	–	–	–	–	CH <sub>2</sub> 3.19(m)/3.55(m)	58.8
–	–	–	–	–	CH <sub>3</sub> 1.16 (t, 6.6)	14.8
17-OCH <sub>3</sub>	3.79 (s)	52.2	3.79 (s)	52.1	–	–

the positive ion peak at  $m/z$  553.2645  $[\text{M} + \text{H}]^+$ . Its IR spectrum exhibited absorption bands at 3570 and 1745  $\text{cm}^{-1}$ , indicating the existence of the hydroxyl and ester carbonyl groups. The presence of an  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone moiety was substantiated by the signals for the olefin proton and carbon signals at  $\delta_{\text{H}}$  6.06 and  $\delta_{\text{C}}$  116.4, respectively, the downfield olefin quaternary carbon at  $\delta_{\text{C}}$  172.5, the lactone carbonyl at  $\delta_{\text{C}}$  168.6, and the hemiketal carbon at  $\delta_{\text{C}}$  106.3. The  $^1\text{H}$  NMR spectrum (Table 1) showed the presence of four tertiary methyl groups at  $\delta_{\text{H}}$  1.13 (3H, s), 1.14 (3H, s), 1.15 (3H, s), 1.57 (3H, s), three acetoxy groups at  $\delta_{\text{H}}$  2.01 (3H, s), 2.14 (3H, s), 2.08 (3H, s), three low-field protons attached to carbon atoms bearing an oxygen function at  $\delta_{\text{H}}$  5.65 (1H, t,  $J = 9.0$  Hz), 5.47 (1H, d,  $J = 8.4$  Hz), 4.87 (1H, br s), and the signals at  $\delta_{\text{H}}$  2.93 (s, C<sub>5</sub>-OH), 2.31 (s, C<sub>14</sub>-OH) for two hydroxyl groups. Except the signals for the  $\alpha,\beta$ -unsaturated  $\gamma$ -lactone moiety and three acetoxy groups, the  $^{13}\text{C}$  NMR spectrum (Table 1) also showed 18 carbon signals for four tertiary methyl groups, six carbon atoms bearing the oxygen atom ( $\delta_{\text{C}}$  79.1, 75.1, 74.9, 74.7, 72.6, 58.8), and other eight alkyl carbons. These  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data were similar to those of neocaesalpin L [9], except for the presence of the ethyl group at  $\delta_{\text{H}}$  1.16 (3H, t, 6.6 Hz), 3.55 (1H, m)/3.19 (1H, m), and  $\delta_{\text{C}}$  14.8, 58.8. This was confirmed by the analysis of the HSQC and HMBC spectra (Figure 2). The location of the CH<sub>3</sub>CH<sub>2</sub>O– group was determined to be at C-12 on the basis of the long-range correlations of the protons at  $\delta_{\text{H}}$  3.55 and 3.19 with the hemiketal carbon at  $\delta_{\text{C}}$  106.3. Moreover, in the NOESY spectrum of compound **3** (Figure 2), the presence of mutual correlations of CH<sub>3</sub>-20/H-6<sub>ax</sub>, H-8, CH<sub>3</sub>-19; CH<sub>3</sub>-19/H-1; H-9/C<sub>5</sub>-OH/C<sub>1</sub>-OCOCH<sub>3</sub>; H-9/CH<sub>3</sub>-17, C<sub>12</sub>-OCH<sub>2</sub>CH<sub>3</sub>, and H-7/CH<sub>3</sub>-17 agreed with the relative structure of neocaesalpin L. Based on the above analysis, **3** was determined as

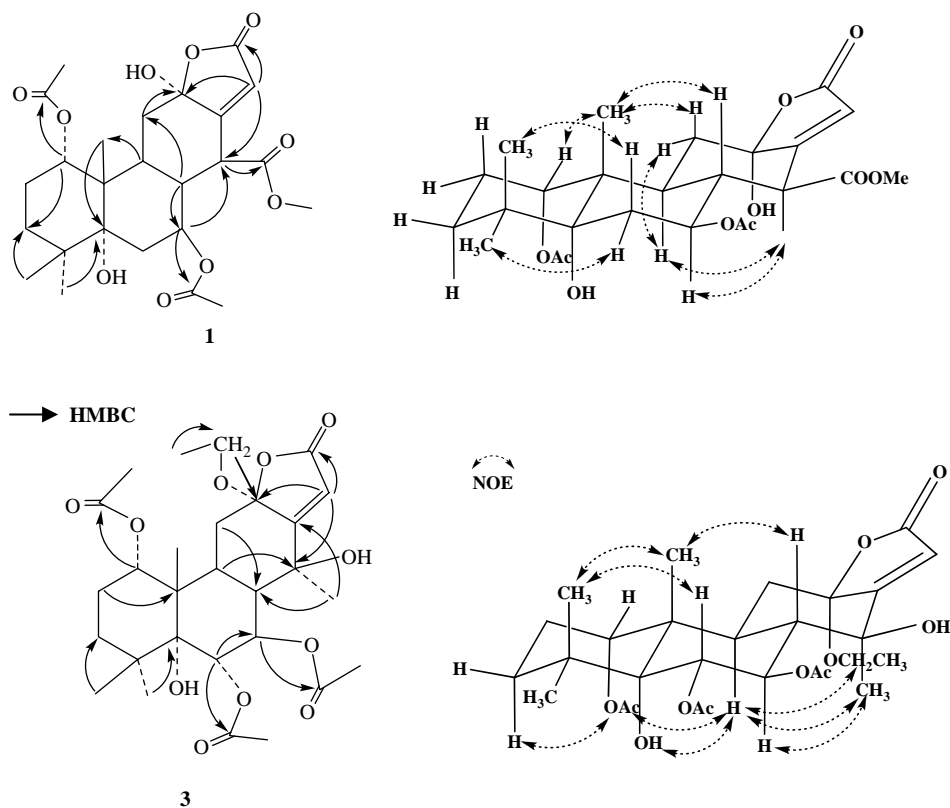


Figure 2. Key HMBC and NOESY correlations of **1** and **3**.

12 $\alpha$ -ethoxyl-1 $\alpha$ ,6 $\alpha$ ,7 $\beta$ -triacetoxy-5 $\alpha$ ,14 $\beta$ -dihydroxy-cass-13(15)-en-16,12-olide.

### 3. Experimental

#### 3.1 General experimental procedures

Optical rotations were measured on a Jasco P-1010 polarimeter. Melting points were determined on a Yanagimoto MP-500D. UV spectra were performed on a UV-1700 spectrophotometer. IR spectra were recorded on KBr discs with a Bruker IFS-55 spectrometer. 1D and 2D NMR spectral data were taken on a Bruker AV-600 spectrometer with TMS as the internal standard. HR-ESI-MS were recorded on a Bruker micro-TOFQ mass spectrometer. HPLC separation was carried out on a reversed phase Mightysil packed column using the gradient CH<sub>3</sub>CN–H<sub>2</sub>O and

MeOH–H<sub>2</sub>O solvent systems with detection at 210 nm. Silica gel for column chromatography (CC, 200–300 mesh) and TLC plates (GF<sub>254</sub>) were purchased from Qingdao Marine Chemical Ltd (Qingdao, China), and spots were visualized by spraying the plates with 10% H<sub>2</sub>SO<sub>4</sub> solution, followed by heating. Sephadex LH-20 was purchased from Pharmacia Biotech (Pharmacia, Kalamazoo, MI, USA). All chemical agents used were of biochemical reagent grade.

#### 3.2 Plant material

The seeds of *C. minax* Hance were purchased from Nanning City, Guangxi Zhuang Autonomous Region of China in March, 2008 and identified by Prof. Qi-shi Sun of the School of Traditional Chinese

Medicine, Shenyang Pharmaceutical University, in which a voucher specimen (No. ZB-2008-017) has been deposited.

### 3.3 Extraction and isolation

Powdered seeds (15 kg) were extracted with 95% ethanol under reflux for three times. The solution was evaporated in vacuum to give a brown viscous residue (1200 g), which was suspended in water and partitioned successively with chloroform, ethyl acetate, and *n*-butanol. The chloroform soluble fraction (175 g) was subjected to silica gel column chromatography, eluted with a chloroform–methanol (100:1–1:1) gradient system, to give 10 fractions (A–J). Fraction B (1.0 g) was subjected to silica gel column chromatography, eluted with a gradient of petroleum ether–acetone (10:1–1:1) to afford three subfractions. Subfraction 2 was then purified by Sephadex LH-20 column chromatography with chloroform–methanol (1:1) and RP-HPLC with MeOH–H<sub>2</sub>O (50:50) to afford **3** (12.2 mg, 0.001%). Fraction D (3.6 g) was subjected to silica gel column chromatography, eluted with a gradient of petroleum ether–acetone (20:1–1:1) to afford five fractions. Subfraction 3 was purified by repeated silica gel column chromatography and preparative TLC with petroleum ether–acetone (3:1) to afford **1** (8.0 mg, 0.00067%) and **2** (3.5 mg, 0.00029%).

#### 3.3.1 Methyl 1 $\alpha$ ,7 $\beta$ -diacetoxy-5 $\alpha$ ,12 $\alpha$ -dihydroxy-cass-13(15)-en-16,12-olide-17 $\beta$ -carboxylate (**1**)

Colorless needles (CDCl<sub>3</sub>),  $[\alpha]_D^{20}$ : –17.3 ( $c = 0.22$ , CHCl<sub>3</sub>); mp 253–255°C; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ): 214 (3.59) nm; IR (KBr, cm<sup>-1</sup>)  $\nu_{\max}$ : 3579, 3408, 2955, 1744, 1367, 1236, 1023, 757; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) spectral data: see Table 1; HR-ESI-MS:  $m/z$  495.2222 [M + H]<sup>+</sup> (calcd for C<sub>25</sub>H<sub>35</sub>O<sub>10</sub>, 495.2225).

#### 3.3.2 Methyl 7 $\beta$ -acetoxy-1 $\alpha$ ,5 $\alpha$ ,12 $\alpha$ -trihydroxy-cass-13(15)-en-16,12-olide-17 $\beta$ -carboxylate (**2**)

White needles (CHCl<sub>3</sub>),  $[\alpha]_D^{20}$ : –29.1 ( $c = 0.12$ , CHCl<sub>3</sub>); mp 277–279°C; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ): 214 (3.57) nm; IR (KBr, cm<sup>-1</sup>)  $\nu_{\max}$ : 3366, 2974, 2925, 1747, 1383, 1217, 1048, 758; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) spectral data: see Table 1; HR-ESI-MS:  $m/z$  475.1941 [M + Na]<sup>+</sup> (calcd for C<sub>23</sub>H<sub>32</sub>O<sub>9</sub>Na, 475.1939).

#### 3.3.3 12 $\alpha$ -Ethoxyl-1 $\alpha$ ,6 $\alpha$ ,7 $\beta$ -triacetoxy-5 $\alpha$ ,14 $\beta$ -dihydroxy-cass-13(15)-en-16,12-olide (**3**)

White amorphous solid,  $[\alpha]_D^{20}$ : –94.1 ( $c = 0.09$ , CHCl<sub>3</sub>); mp 239–241°C; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ): 216 (3.29) nm; IR (KBr, cm<sup>-1</sup>)  $\nu_{\max}$ : 3570, 2981, 2936, 1745, 1373, 1232, 1033, 756; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) spectral data: see Table 1; HR-ESI-MS:  $m/z$  553.2645 [M + H]<sup>+</sup> (calcd for C<sub>28</sub>H<sub>41</sub>O<sub>11</sub>, 553.2643).

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